

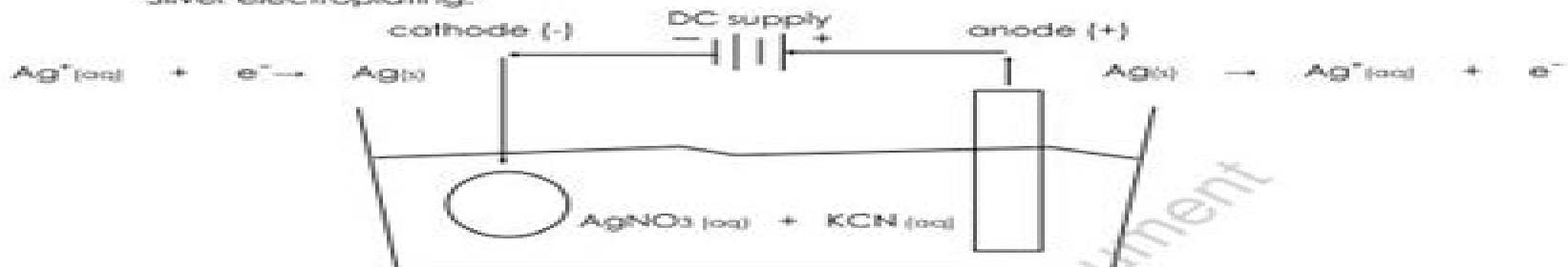
19. Electroplating

Objects may be plated with silver or gold for aesthetic incentive

The object to be plated is always placed at the cathode

The material to provide the plating is always placed at the anode

Silver electroplating:



The mass of the silver electrode at the anode decreases over time as a thin silver layer forms on the object at the cathode

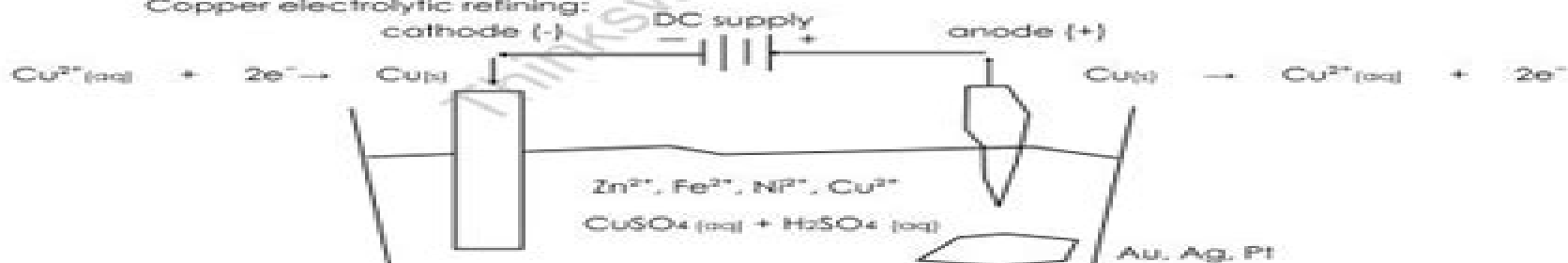
Tin plating may also be done through this method

20. Electrolytic purification

Copper and gold can both be purified through electrolysis

Impure sample ["blister copper"] placed at the anode and a pure sample forms at the cathode

Copper electrolytic refining:



DC supply is small so only the metals more reactive than copper will oxidise. Thus, gold, silver and platinum components of the impure sample fall to the bottom of the cell in a 'anode sludge' as they will only oxidise with a higher DC voltage.

All other reactive metals oxidise forming ions but only copper ions will be able to reduce at the cathode as all other metals will require a higher DC voltage.

The cathode will hence become a 99.95% pure sample of copper

Dot Point Wace Chemistry

**William GREGORY (Professor of
Chemistry in the University of
Edinburgh.)**



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Dot Point WACE Chemistry Units 2A And 2B Marilyn Schell,Margaret Hogan,Peter Wong,2009-01-01 The Nation,1877 **The American Bookseller** ,1891 **A New English Dictionary on Historical Principles** James Augustus Henry Murray,Sir William Alexander Craigie,Charles Talbut Onions,1909 Science Citation Index ,1975 Vols for 1964 have guides and journal lists **Dot Point IB Chemistry AHL** Marilyn Schell,Margaret Hogan,2010 The Compact Edition of the Oxford English Dictionary Sir James Augustus Henry Murray,1971 Micrographic reproduction of the 13 volume Oxford English dictionary published in 1933 Chemistry Mike Lucarelli,David V. Proctor,2009 **Chemistry** Mike Lucarelli,David V. Proctor,2008 WACE Chemistry Units 3&4 Complete Course Notes ,2022-03 **Chemistry Unit 4** Jack Anderson,Georgia Anderson,Tyler Clement,Aryan Puri,Rabea Zeynali,2021-05 ThrivEd is a student run not for profit organisation that writes textbooks on the philosophy of writing textbooks that truly empower students Our Unit 4 Chemistry textbook is a 224 page textbook that covers Organic Chemistry and Chemical Synthesis The proceeds from every two books sold are used to produce and donate a third textbook for a student who would benefit from the same support The book is structured in five main sections 1 Bridging the Gap To create the connections needed to set off on the right foot in Year 12 the Bridging the Gap sections are designed to provide in depth teachings of the foundational Year 11 content 2 Main Content By extensively working through concepts with no assumptions and teaching with clear explanations worked examples and real world applications we hope to make each chapter an exciting empowering and well connected experience 3 Topic Test and Exam Notes To guide your preparation of personal notes leading up to assessments sample test and exam notes have been provided to make this process effective and intentional 4 Progressive Repetitive and Advanced Questions Each progressive question is designed to progressively teach each concept and is paired with multiple repetitive questions allowing you to stop and practice anything you like Advanced questions are also there for some extra fun 5 Online Answer Guide With each question comes a three stage answer including a sample response points to note about the question and a WACE inspired marking guide allowing each question to be understood in all its elements These answer guides are free as a downloadable PDFs It is with this structure that the Unit is extensively covered and targeted to provide the greatest possible benefits to students **Dot Point** Marilyn Schell,Margaret Hogan,2018 **VCE Chemistry** Marilyn Schell,Margaret Hogan,2017 Master the syllabus content point by point Exam style questions on all syllabus dot points Write your answers in the spaces provided Check your responses with the answers supplied Back cover Essential Concepts in Chemistry and Zumdahl Interactive CD-ROM Four Point Zero Sherman,1999-01-01 Chemistry for Western Australia Cherie Lewis,Peter Lewis,2008 Chemistry for Stage 2 Units 2A and 2B Provided by publisher **A Study Guide to Chemistry** Julia Burdge,McGraw-Hill Higher Education,Raymond Chang,2008-02-01 Chemistry by Discovery G. van Praagh,1964 Chemistry 12 Castle Rock Research Corp,2006-01-01 **A Fourteen Weeks Course in Chemistry** Joel Dorman

Steele, 1869 **Outlines of Chemistry, for the use of students** William GREGORY (Professor of Chemistry in the University of Edinburgh.), 1847

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