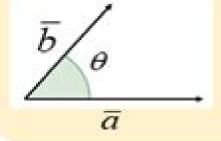
# **Dot Product**

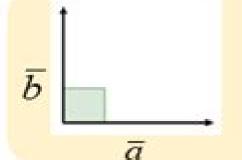
If  $\overline{a}=< a_1,a_2,a_3>$  and  $\overline{b}=< b_1,b_2,b_3>$  then the dot product is

$$\overline{a} \bullet \overline{b} = a_1 b_1 + a_2 b_2 + a_3 b_3$$



If  $\theta$  is the angle between  $\bar{a}$  and  $\bar{b}$  then

$$\overline{a} \bullet \overline{b} = |\overline{a}| |\overline{b}| |\cos \theta|$$



 $\overline{a} \bullet \overline{b}$  are orthogonal (perpendicular) if and only if  $\overline{a} \bullet \overline{b} = 0$ 

# **Dot Product Math Is Fun**

Ray Toal, Sage Strieker, Marco Berardini

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simulation to help solve basic to more advanced problems Michael W Roth PhD serves as Dean of the School of STEM and Business at Hawkeye Community College in Waterloo Iowa He was most recently Chair for three years at Northern Kentucky University's Department of Physics Geology and Engineering Technology and holds several awards for teaching excellence

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web gather data use the gizmo to measure the solubility of potassium nitrate at each temperature given in the table below then graph the resulting solubility curve at right temperature solubility g 100 ml 10 c 21g 100ml 20 c 31g 100ml 30 c 44g 100ml 40 c 62g 100ml 50 c 83g 100ml 60 c 108g 100ml 70 c 136g 100ml 80 c 168g

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water gizmo w arm up a solution generally consists of two parts a solute that is dissolved and a solvent that the solute is dissolved into for example sugar is a solute that is dissolved into the solvent water in the

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web 1 in this solution what is the solute potassium nitrate what is the solvent water 2 click add 10 g to mix 10 g of potassium nitrate into the water a did all of the potassium nitrate dissolve b how can you tell yes all of the potassium nitrate disappeared as it was stirred into the water activity a solubility

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