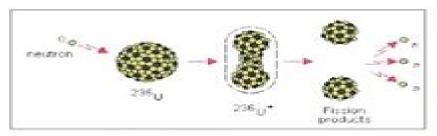
Nuclear Fission vs. Nuclear Fusion

Nuclear f "i" s s i O n (Spi"i"tting an atom into two new ones)

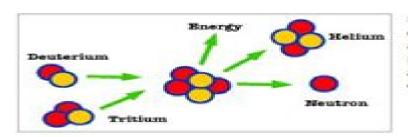


In nuclear fission reactions (also called radioactive decay), a neutron is aimed at the nucleus of a large, unstable atom, like uranium, thorium, or other radioactive elements. The extra mass of the neutron causes the radioactive nucleus to split apart, forming lighter elements, free neutrons, and great quantities of energy. This process causes convection currents that move Earth's tectonic plates, and generate earthquakes and volcanic eruptions.

Nuclear Fission:

- Source of energy in the core of the Earth that produces heat from the decay of radioactive elements.
- · Produces vast quantities of energy.
- Does not produce particulate air pollution like fossil fuels and coal.
- Involves the splitting of harmful radioactive elements.
- Loss of control leads to harmful radiation exposure.
- Produces a radioactive waste product that will need to be stored.

Nuclear f "u" s i o n (combining/f"u"sing two atoms together into one new atom)



During fusion reactions, nuclei collide and fuse, eventually forming nuclei of heavier elements and producing enormous amounts of energy. Fusion of hydrogen to helium occurs in the sun and is the source for all external weather related events.

Nuclear Fusion:

- · Source of energy in the Sun that produces heat from the fusing of elements like hydrogen.
- · Produces unsurpassed quantities of energy.
- Does not produce particulate air pollution like fossil fuels and coal.
- Does not produce a radioactive waste product that will need to be stored.
- Currently, we lack the technology to maintain reactions as a viable energy source.

Fission Fusion Worksheet Nuclear Weapons Answers

Yan Bai

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