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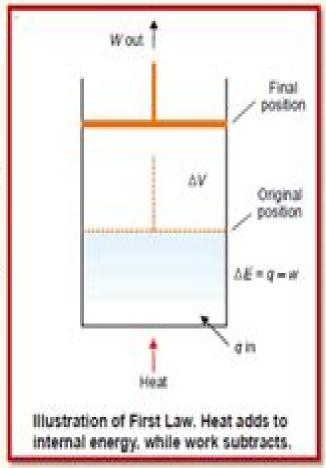
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First law of Thermodynamics + solved problems

Different definitions of First Law of Thermodynamics

 $\Delta E = q - w$

- (1) The total energy of an isolated system remains constant though it may change from one form to another.
- (2) Whenever energy of a particular type disappears equivalent amount of another type must be produced.
- (3) Total energy of a system and surroundings remains constant (or conserved)
- (4) It is impossible to construct a perpetual motion machine that can produce work without spending energy on it.



First Law Of Thermodynamics Problems And Solutions

RS Peters

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