

Concept Review

Section: Families of Elements

1. **Classify** each of the following elements as an alkali metal, alkaline-earth metal, transition metal, or semiconductor based on its position in the periodic table.

_____ **a.** rubidium, Rb
_____ **b.** silicon, Si
_____ **c.** silver, Ag
_____ **d.** barium, Ba

2. **Classify** each of the following elements as a halogen, noble gas, or other nonmetal based on its position in the periodic table.

_____ **a.** carbon, C
_____ **b.** chlorine, Cl
_____ **c.** radon, Rn
_____ **d.** phosphorus, P

3. **Predict** which of the following ions would be likely to form:

_____ a. Na^{2+}	_____ d. Br^-
_____ b. Cl^+	_____ e. Ne^-
_____ c. Ca^{2+}	_____ f. Ne^+

4. **Explain** why chlorine, Cl, is very reactive, whereas argon, Ar, is unreactive.

5. **Analyze** the following pairs of elements, and determine whether each pair has similar or different reactivities.

_____ **a.** potassium, K, and rubidium, Rb
_____ **b.** calcium, Ca, and barium, Ba
_____ **c.** sodium, Na, and chlorine, Cl
_____ **d.** helium, He, and krypton, Kr

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**A. J. Mooradian, Atomic Energy of
Canada, Chalk River Nuclear
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