

Chapter 1

Introduction to Electronics

Question 1.1: Atomic Structure

- 1. The atom is the smallest constituent of matter that cannot be created or destroyed.
- 2. The atom consists of three subparticles, viz. electron, proton and neutron.

Question 1.2: Materials Used for Electronics

- 1. The materials are generally classified into four categories: (a) conductors, (b) semiconductors, (c) insulators and (d) superconductors.
- 2. The atoms containing valence electrons are called conductors.
- 3. An atom in which all the valence electrons are paired is called an insulator.

Question 1.3: Current in Semiconductors

- 1. Current is called as drift current when the electrons and holes are pushed.
- 2. Current is produced when there is a potential difference across the material.

Question 1.4: N-Type and P-Type Semiconductors

- 1. In an N-type semiconductor, the donor atoms are present in pure silicon. These atoms have five valence electrons. When one of these atoms is replaced by a silicon atom, it becomes an N-type semiconductor.
- 2. In a P-type semiconductor, the acceptor atoms are present in pure silicon. These atoms have three valence electrons. When one of these atoms is replaced by a silicon atom, it becomes a P-type semiconductor.

Question 1.5: The PN Junction

- 1. The PN junction is formed by joining an N-type semiconductor with a P-type semiconductor. The junction is formed by the diffusion of electrons from the N-type semiconductor to the P-type semiconductor. The junction is formed by the diffusion of holes from the P-type semiconductor to the N-type semiconductor.
- 2. The PN junction is a device which is used to convert electrical energy into light energy. It is used in many applications, such as in the construction of diodes, transistors, and integrated circuits.

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Michael Seilmaier



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