Quiz 3 for Physics 176: Answers

Professor Greenside

True or False Questions (2 points each)

For each of the following statements, please circle T or F to indicate respectively whether a given statement is true or false.

 T / F If the volume V of an ideal monostomic gas is halved during an isothermal compression that conserves the number of particles, then the entropy of the gas will decrease by Nk ln(2).

Answer: T This was discussed in Schroeder and in lecture and is a direct consequence of the Sackur-Tetrode equation, which was given on the first page of the quiz. If the number of particles Nand total particle energy U are fixed, then Sackur-Tetrode says that $S = Nk \ln(V) + f(N, U)$, where fis a function that depends only on N and U and so does not change during a volume change and so can be ignored when computing the entropy change. This problem needed to assume an isothermal compression to prevent the energy U from changing during the compression.

Thus the change in entropy during the isothermal compression of an ideal monoatomic gas is $\Delta S = S_f - S_i = Nk \ln(V_f) - Nk \ln(V_i) = Nk \ln(V_f/V_i)$. For this problem, $V_f/V_i = 1/2$ so the change in entropy is $\Delta S = Nk \ln(1/2) = -Nk \ln(2)$, which is indeed a decrease in the entropy by the magnitude $Nk \ln(2) = 0.7Nk$.

 T / F The product of an extensive thermodynamic quantity with an intensive thermodynamic quantity is also an extensive quantity.

Answer: T This question was intended to test whether you understood my brief description in class about extensive variables like M, V, S, and U versus intensive variables like P, T, $\rho = M/V$, n = N/V, and so on. Extensive variables increase in proportion to the volume of the system (keeping the density N/V constant), e.g. if you double the volume you typically double the entropy and the mass. Intensive variables are almost always ratios of two extensive variables and so don't change when the volume is varied. From this, you should be able to see that the product of an extensive variable with an intensive variable remains extensive: changing the volume causes the extensive variable to change, but the intensive factor does not change.

T / F Given that the function f(x) = 1 − x⁴ has a single maximum at x_{max} = 0 with value f_{max} = 1, the function [f(x)]^N for N a large integer can be accurately approximated by a Gaussian e^{-N,5x²} for A > 0 some constant.

Answer: F This question was intended to see if you could think rather than just memorize. In class, we discussed a "theorem", that a function f(x) with a unique global maximum, if raised to a large power N, $f(x)^N$, behaves like a narrow Gaussian function centered on the location x_{max} of the global maximum. The derivation involved taking the log of the function $f(x)^N$ to get $N \log(f(x))$. Taylor expanding f(x) about $x = x_{max}$, and using our favorite log approximation $\log(a + \epsilon) = \log(a) + \epsilon/a$.

A key step in the derivation was taking the Taylor expansion to second order:

$$f(x) = f(x_{max}) + f'(x_{max})(x - x_{max}) + \frac{1}{2!}f''(x_{max})(x - x_{max})^2 + ...$$
 (1)

$$\approx f_{\text{max}} + \frac{f''(x_{\text{max}})}{2} (x - x_{\text{max}})^2. \qquad (2)$$

But this procedure only works if the Taylor expansion has a nonzero coefficient $f''(x_{max})$ at the location of the global maximum, otherwise you get some other kind of functional behavior that is not a Gaussian.

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Table of Contents Entropy Question And Answers Download

- 1. Understanding the eBook Entropy Question And Answers Download
 - The Rise of Digital Reading Entropy Question And Answers Download
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Entropy Question And Answers Download
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Entropy Question And Answers Download
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Entropy Question And Answers Download
 - Personalized Recommendations
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- 8. Staying Engaged with Entropy Question And Answers Download
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Entropy Question And Answers Download
- 9. Balancing eBooks and Physical Books Entropy Question And Answers Download
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Entropy Question And Answers Download
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Entropy Question And Answers Download
 - Setting Reading Goals Entropy Question And Answers Download
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Entropy Question And Answers Download
 - Fact-Checking eBook Content of Entropy Question And Answers Download
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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