

Sample results. Actual results may vary.

PATIENT INFORMATION

REPORT STATUS: FINAL

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SPECIMEN INFORMATION

SPECIMEN:
REQUISITION:
LAB REF NO:

DOB:
AGE:
GENDER:
FASTING:

Clinical Info:

COLLECTED:
RECEIVED:
REPORTED:

Test Name	Result	Flag	Reference Range	Lab
DRUG ABUSE PANEL 10-50 NO CONFIRM				EN
PLEASE NOTE:				EN
* These results are for medical treatment only. *				
* Analysis was performed as non-forensic testing. *				
AMPHETAMINES (1000 ng/mL SCREEN)	NEGATIVE			EN
BARIATURATES	NEGATIVE			EN
BENZODIAZEPINES	NEGATIVE			EN
COCAINE METABOLITES	NEGATIVE			EN
MARIJUANA METABOLITES (50 ng/mL)	NEGATIVE			EN
METHADONE	NEGATIVE			EN
METHAQUALONE	NEGATIVE			EN
OPIATES	NEGATIVE			EN
PERICYCLIDINE	NEGATIVE			EN
PROPOXYFENE	NEGATIVE			EN
PLEASE NOTE:	DSR			EN
COMMENT				EN

THE SUBMITTED URINE SPECIMEN WAS TESTED AT THE CUTOFFS LISTED BELOW.

DRUG CLASS	INITIAL TEST CUTOFF
AMPHETAMINES	1000 ng/mL
BARIATURATES	300 ng/mL
BENZODIAZEPINES	300 ng/mL
COCAINE METABOLITES	300 ng/mL
MARIJUANA METABOLITES	50 ng/mL
METHADONE	300 ng/mL
METHAQUALONE	300 ng/mL
OPIATES	300 ng/mL
PERICYCLIDINE	25 ng/mL
PROPOXYFENE	300 ng/mL

PLEASE READ THIS IMPORTANT MESSAGE:

THIS DRUG SCREEN IS FOR MEDICAL USE ONLY. THE RESULTS ARE PRESUMPTIVE; BASED ONLY ON SCREENING METHODS, AND THEY HAVE NOT BEEN CONFIRMED BY A SECOND INDEPENDENT CHEMICAL METHOD. THESE RESULTS SHOULD BE USED ONLY BY PHYSICIANS TO RENDER DIAGNOSIS OF TREATMENT, OR TO MONITOR PROGRESS OF MEDICAL CONDITIONS.

Performing Laboratory Information:

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web we then changed the volume between 5ml and 20ml and recorded the pressure at each this part of the lab showed that volume and pressure are inversely proportionate as the pressure would decrease if the volume was increased part b of the experiment looked at the relationship between temperature and pressure

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of

gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $pV = nRT$ is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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web the pressure volume temperature pvt fluids lab in ku s chemical petroleum engineering department is used to study the properties of fluids under a wide range of pressure and temperature

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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