

First Name _____ Last Name (Surname) _____ **Key**

1. (16 points) Using the reduction potentials provided, circle the appropriate choice in each case.

4 pts
each

Strongest Reducing Agent: VO^{2+} or Cr^{2+} (circle one)



See explanation
below

Strongest Oxidizing Agent: Mn^{2+} or Al^{3+} (circle one)



Strongest Reducing Agent: Cr or Zn (circle one)



Strongest Oxidizing Agent: Au^{3+} or Ag^+ (circle one)



2. (6 points) Determine the oxidation number in each case.



2 pts
each

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M Mosston



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