

First Name \_\_\_\_\_

Last Name (Surname) \_\_\_\_\_

Key

1. (16 points) Using the reduction potentials provided, circle the appropriate choice in each case.

*4 pts  
each*

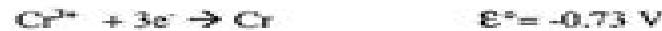
Strongest Reducing Agent:  $\text{VO}^{2+}$  or  $\text{Cr}^{2+}$  (circle one)

*See explanation  
below*

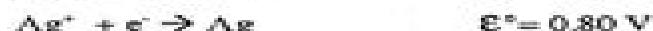
Strongest Oxidizing Agent:  $\text{Mn}^{2+}$  or  $\text{Al}^{3+}$  (circle one)



Strongest Reducing Agent:  $\text{Cr}$  or  $\text{Zn}$  (circle one)



Strongest Oxidizing Agent:  $\text{Au}^{3+}$  or  $\text{Ag}^+$  (circle one)



2. (6 points) Determine the oxidation number in each case.

*2 pts  
each*

# College Chemistry Questions And Answers

**M Mosston**

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