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# CLASSIFYING CHEMICAL REACTIONS

Classify the following chemical reactions as one of the following: synthesis, decomposition, single-replacement, double-replacement, or combustion.

$2H_0 + O_0 \rightarrow 2H_0O$	caco, cao + co,
3K + 3H <sub>2</sub> O → H <sub>4</sub> + 3KOH	Caco, • 2HCI Caci, • H <sub>2</sub> CO,
290010 <sub>9</sub> 28001 + 30 <sub>9</sub>	Ca + 2HCl CaCl <sub>2</sub> + H <sub>2</sub>
280 <sub>e</sub> + 0 <sub>e</sub> 280 <sub>e</sub>	PEONOGY = 2HCI → PECI, = 2HHO,
Ou + 2AgNO, $\rightarrow$ 2Ag + Ou040 $\phi_0$	294a • Cl <sub>e</sub> 25eeCli
2H <sub>2</sub> O 2H <sub>2</sub> - O <sub>2</sub>	OH, + 20, 00, + 24,0
$PbONO_{\mathcal{O}_{S}} = 290  \rightarrow  PbI_{b} = 29000_{b}$	H,CO, CO, - H,O
2Fe + O₂ → 2FeO	Zn + 2HCl H <sub>6</sub> + ZnCl <sub>6</sub>
HCI + NaOH → NaOI + HOH	25% + 254,0 25%0H + H <sub>2</sub>

## **Get The Message Classifying Chemical Reactions**

**Joan Tomsic** 

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