

Name: [KEY]

Period: \_\_\_\_\_ Date: \_\_\_\_\_

**I. Dilution Problems**  $M_1V_1 = M_2V_2$ 

1. A student wishes to prepare 250. mL of a 0.15 M NaOH solution from a 6.00M NaOH stock solution. What volume of the stock solution should she use?

$$\begin{array}{l} M_1 = 0.15 \text{ M} \quad M_1 = 6.00 \text{ M} \\ V_1 = ? \quad V_2 = 250. \text{ mL} \end{array} \quad (6.00 \text{ M})(x \text{ mL}) = (0.15 \text{ M})(250. \text{ mL})$$

$$x = 16.25 \text{ mL}$$

+ 6.25 mL stock solution

2. How much water must be added to a 2.50 M HCl solution to obtain 500. mL of a 0.100M HCl solution?

$$\begin{array}{l} M_1 = 2.50 \text{ M} \quad M_2 = 0.100 \text{ M} \\ V_1 = ? \quad V_2 = 500. \text{ mL} \end{array} \quad (2.50 \text{ M})(x \text{ mL}) = (0.100 \text{ M})(500. \text{ mL})$$

$$x = 20 \text{ mL}$$

$$\frac{500 \text{ mL}}{\text{(total volume)}} - \frac{20 \text{ mL}}{\text{(stock solution)}} = 480 \text{ mL}$$

Volume of water added

3. What is the final volume of a 0.15 M solution prepared from 25.0 mL of a 6.0 M sodium acetate solution?

$$\begin{array}{l} M_1 = 6.0 \text{ M} \quad M_2 = 0.15 \text{ M} \\ V_1 = 25.0 \text{ mL} \quad V_2 = ? \end{array} \quad (6.0 \text{ M})(25.0 \text{ mL}) = (0.15 \text{ M})(x \text{ mL})$$

$$x = 1000 \text{ mL}$$

**II. Molarity Problems**  $\text{Molarity (M)} = \frac{\text{mol}}{\text{L}}$ 

4. What is the molarity of a solution that contains 12.5g NaC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> (Sodium Acetate) dissolved in 0.250 L?

$$12.5 \text{ g NaC}_2\text{H}_3\text{O}_2 \times \frac{1 \text{ mole of NaC}_2\text{H}_3\text{O}_2}{82.04 \text{ g NaC}_2\text{H}_3\text{O}_2} = 0.1524 \text{ moles}$$

$$M = \frac{0.1524 \text{ mol}}{0.250 \text{ L}} = 0.609 \text{ M}$$

5. What mass of citric acid (molar mass = 192.14 g/mol) would be needed to make 0.10 L of 5.00 M citric acid solution?

$$5.00 \text{ M} = \frac{x \text{ moles}}{0.10 \text{ L}} \quad 0.50 \text{ moles citric acid} \times \frac{192.14 \text{ g citric acid}}{1 \text{ mol citric acid}} = 96.07 \text{ g citric acid}$$

$$x = 0.50 \text{ moles}$$

96.1 g of citric acid

6. What mass of sodium hydroxide is needed to make 0.500 L of 2.50 M NaOH?

$$2.50 \text{ M} = \frac{x \text{ moles}}{0.500 \text{ L}} \quad 1.25 \text{ moles NaOH} \times \frac{40 \text{ g NaOH}}{1 \text{ mol NaOH}} = 50 \text{ g of NaOH}$$

$$x = 1.25 \text{ moles}$$

# Dilutions Cavalcade 2004

**Yijin Wang**

## Dilutions Cavalcade 2004:

**Mergent Corporate News Reports Monthly**,2004-05     **A Colossal Wreck** Alexander Cockburn,2013-09-10  
Alexander Cockburn was without question one of the most influential journalists of his generation whose writing stems from the best tradition of Mark Twain H L Mencken and Tom Paine Colossal Wreck his final work finished shortly before his death in July 2012 exemplifies the prodigious literary brio that made Cockburn s name Whether ruthlessly exposing Beltway hypocrisy pricking the pomposity of those in power or tirelessly defending the rights of the oppressed Cockburn never pulled his punches and always landed a blow where it mattered In this panoramic work covering nearly two decades of American culture and politics he explores subjects as varied as the sex life of Bill Clinton and the best way to cook wild turkey He stands up for the rights of prisoners on death row and exposes the chicanery of the media and the duplicity of the political elite As he pursues a serpentine path through the nation he charts the fortunes of friends famous relatives and sworn enemies alike to hilarious effect This is a thrilling trip through the reefs and shoals of politics and everyday life Combining a passion for the places the food and the people he encountered on dozens of cross country journeys Cockburn reports back over seventeen years of tumultuous change among what he affectionately called the thousand landscapes of the United States     *Journeys of Desire* Alastair Phillips,Ginette Vincendeau,2019-07-25 A comprehensive guide to European actors in American film this book brings together 15 chapters with A Z entries on over 900 individuals It includes case studies of prominent individuals and phenomena associated with the emigres such as the stereotyping of European actresses in bad women roles and the irony of Jewish actors playing Nazis     *Mergent Company Archives Manual* ,2002 Contains the final statistical record of companies which merged were acquired went bankrupt or otherwise disappeared as private companies

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