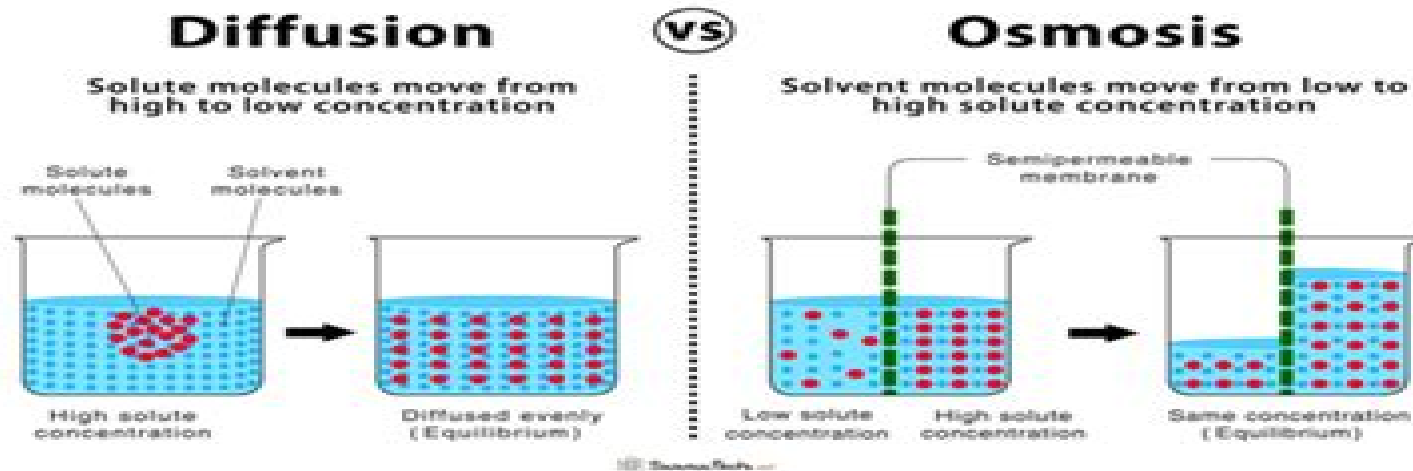


## Diffusion and Osmosis

### Background:

**Diffusion** is the passive movement of particles of a solute from an area of high concentration to an area of low concentration.

**Osmosis** is the passive movement of a solvent, such as water, through a semipermeable membrane from a region of high solvent concentration to an area of low solvent concentration.



**Solute:** a substance that has been dissolved in a given solution.

**Solvent:** a substance, usually a liquid, that is capable of dissolving or dispersing one or more other substances.

Ex: To create saltwater, salt is dissolved into water.

Solute: salt

Solvent: water

**Semipermeable membrane:** A membrane or barrier that allows only certain molecules or substances to pass through it and does not allow others.

Ex: Dialysis tubing acts as a semipermeable membrane.

**Lugol's iodine:** a solution of potassium iodide that turns blue in the presence of starch.

# Diffusion Osmosis Lab Manual Answers

**Rajesh Kumar**



## **Diffusion Osmosis Lab Manual Answers:**

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